

## 10. Molecular Dynamics Simulations

# Molecular dynamics simulations

- ▶ a technique for computing equilibrium and transport properties of a classical many-body system
- ▶ very similar to real experiments in many aspects
- ▶ involves preparing a system (a sample), equilibrating it (wait), and computing observables (measurement).
- ▶ common mistakes are very similar to those made in real experiments, such as: the system is not prepared correctly, the measurement is too short, or we are not measuring the right thing.

# Outline

Initialization

Integrating the equation of motion

Computing observables

# Receipe for molecular dynamics simulations

- ▶ common workflow of a molecular dynamics simulation:
  1. initialize the system including the positions and velocities of the particles
  2. run the simulation by integrating the equations of motion
  3. compute the observables of interest
- ▶ all three steps are crucial for a successful simulation
- ▶ the second step is often the most time-consuming part of a simulation

## Initialization

- ▶ assign the initial positions and velocities for all particles
- ▶ initial positions should be compatible with the structure that we want to simulate
- ▶ should not have particles overlapping with each other, which is often achieved by initially placing the particles on a cubic lattice
- ▶ for molecules such as proteins with known structures, we could use the known structure as the initial configuration

## Initialization

- ▶ initial velocities need to be compatible with the temperature of the system
- ▶ from statistical mechanics, we know that the average kinetic energy per degree of freedom is given by

$$\left\langle \frac{1}{2} m v_{\alpha}^2 \right\rangle = \frac{1}{2} k_B T$$

- ▶ more specifically, the velocity components of a particle follow the Boltzmann distribution

$$p(v_{\alpha}) = \frac{1}{\sqrt{2\pi k_B T/m}} e^{-m v_{\alpha}^2 / 2k_B T}$$

- ▶ we can assign the initial velocities by sampling from the Boltzmann distribution
- ▶ what is the average speed of a water molecule at room temperature?

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## The equation of motion

- ▶ the equation of motion is given by Newton's second law

$$m_i \frac{d^2 \mathbf{r}_i}{dt^2} = -\nabla_i U(\mathbf{r}_1, \mathbf{r}_2, \dots, \mathbf{r}_N)$$

- ▶ an equivalent form is the Hamiltonian equations of motion

$$\frac{d\mathbf{q}_i}{dt} = \frac{\partial H}{\partial \mathbf{p}_i}, \quad \frac{d\mathbf{p}_i}{dt} = -\frac{\partial H}{\partial \mathbf{q}_i}$$

where  $\mathbf{q}_i = \mathbf{r}_i$ ,  $\mathbf{p}_i = m_i \mathbf{v}_i$ , and

$$\begin{aligned} H(\mathbf{q}_1, \mathbf{q}_2, \dots, \mathbf{q}_N, \mathbf{p}_1, \mathbf{p}_2, \dots, \mathbf{p}_N) &= K(\mathbf{p}_1, \mathbf{p}_2, \dots, \mathbf{p}_N) + U(\mathbf{q}_1, \mathbf{q}_2, \dots, \mathbf{q}_N) \\ &= \sum_i \frac{\mathbf{p}_i^2}{2m_i} + U(\mathbf{q}_1, \mathbf{q}_2, \dots, \mathbf{q}_N) \end{aligned}$$



## The Hamiltonian equations of motion

- ▶ Hamiltonian equations

$$\begin{aligned}\frac{d\mathbf{q}_i}{dt} &= \frac{\partial H}{\partial \mathbf{p}_i} = \frac{\mathbf{p}_i}{m_i} \\ \frac{d\mathbf{p}_i}{dt} &= -\frac{\partial H}{\partial \mathbf{q}_i} = -\nabla_i U(\mathbf{q}_1, \mathbf{q}_2, \dots, \mathbf{q}_N)\end{aligned}$$

- ▶ in vector form, we have

$$\frac{d}{dt} \begin{pmatrix} \mathbf{q} \\ \mathbf{p} \end{pmatrix} = \begin{pmatrix} 0 & \mathbf{I} \\ -\mathbf{I} & 0 \end{pmatrix} \begin{pmatrix} \frac{\partial H}{\partial \mathbf{q}} \\ \frac{\partial H}{\partial \mathbf{p}} \end{pmatrix}$$

- ▶ a set of first-order differential equations and equivalent to Newton's second law
- ▶ provides a more useful framework for understanding the dynamics of a system

## Integrating the Hamiltonian equations of motion

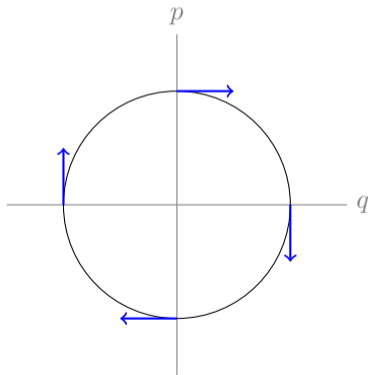
- ▶ a solution of the Hamiltonian equations of motion is a trajectory in the phase space of  $(\mathbf{q}, \mathbf{p})$
- ▶ example: a harmonic oscillator

$$H(q, p) = \frac{p^2}{2m} + \frac{1}{2}kq^2$$

when  $m = 1$  and  $k = 1$

$$\frac{dq}{dt} = \frac{\partial H}{\partial p} = p$$

$$\frac{dp}{dt} = -\frac{\partial H}{\partial q} = -q$$



## Integrating the Hamiltonian equations of motion

- ▶ most Hamiltonian systems are not solvable analytically
- ▶ need numerical methods to integrate the Hamiltonian equations of motion
- ▶ the naive method is the Euler method, which is not a good choice
- ▶ a much better method is the leap-frog Verlet method

## The Euler method

- ▶ naive Euler method

$$p(t + \epsilon) = p(t) + \epsilon \frac{dp}{dt} = p(t) - \epsilon \frac{\partial U}{\partial q}(q(t))$$

$$q(t + \epsilon) = q(t) + \epsilon \frac{dq}{dt} = q(t) + \epsilon \frac{p(t)}{m}$$

- ▶ modified Euler method

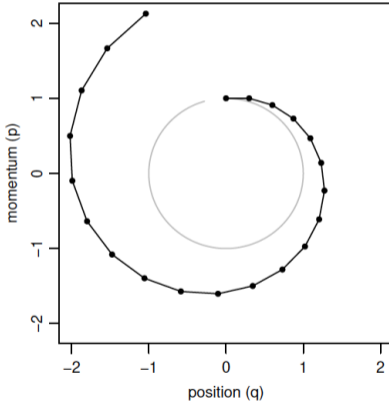
$$p(t + \epsilon) = p(t) - \epsilon \frac{\partial U}{\partial q}(q(t))$$

$$q(t + \epsilon) = q(t) + \epsilon \frac{p(t + \epsilon)}{m}$$

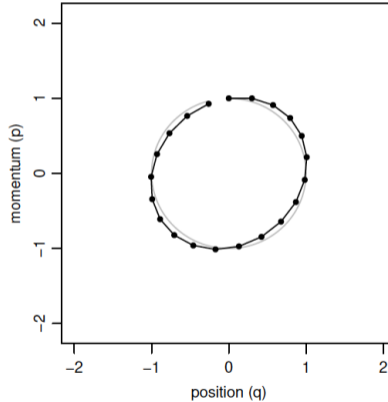
# The Euler method

- ▶ numerical trajectory for the 1-d harmonic oscillator

(a) Euler's Method, stepsize 0.3



(b) Modified Euler's Method, stepsize 0.3



## The leap-frog Verlet method

- ▶ integration scheme

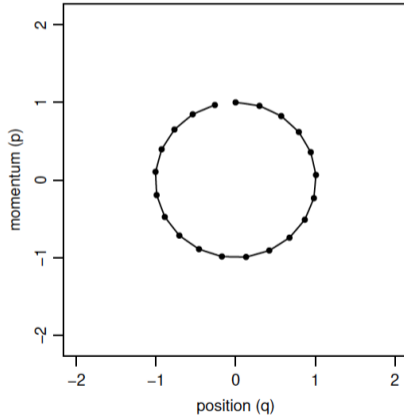
$$p(t + \epsilon/2) = p(t - \epsilon/2) - 2\frac{\partial U}{\partial q}(q(t))$$
$$q(t + \epsilon) = q(t) + \epsilon\frac{p(t + \epsilon/2)}{m}$$

- ▶ is called the leap-frog method because the position and momentum are updated in a staggered fashion
- ▶ is a symplectic integrator which preserves the phase space volume
- ▶ has better energy conservation properties than the Euler method

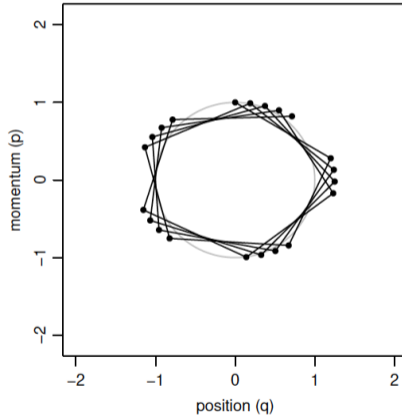
# The leap-frog Verlet method

- ▶ numerical trajectory for the 1-d harmonic oscillator

(c) Leapfrog Method, stepsize 0.3



(d) Leapfrog Method, stepsize 1.2



## The Hamiltonian equations are energy conserving

- ▶ both the Newtonian and Hamiltonian equations of motion are energy conserving
- ▶ trajectories are confined to a constant energy surface in phase space
- ▶ conformations of the system are sampled from the *NVE* ensemble, which is also called the *microcanonical ensemble*
- ▶ we are mostly interested in the *NVT* ensemble, where the temperature is kept constant
- ▶ in the *NVT* ensemble, the system is contacted with and exchange energy with a heat bath
- ▶ in the *NVT* ensemble, the total energy is not conserved but fluctuates
- ▶ integrating the Hamiltonian equations of motion alone will not give us the *NVT* ensemble



# Integrators for the NVT ensemble

- ▶ several integrators are available for the NVT ensemble
  - Andersen thermostat
  - Nose-Hoover thermostat
  - Langevin dynamics
  
- ▶ we will only discuss the Langevin dynamics, which is the most widely used method

## Langevin dynamics

- ▶ the Langevin equation for the  $NVT$  ensemble is given by

$$m_i \frac{d\mathbf{v}_i}{dt} = -\nabla_i U(\mathbf{r}_1, \mathbf{r}_2, \dots, \mathbf{r}_N) - \gamma m_i \mathbf{v}_i + \sqrt{2m_i \gamma k_B T} R_i(t)$$

- ▶  $\gamma$  is the friction coefficient
- ▶  $R_i(t)$  is an uncorrelated random force with zero mean and unit variance
- ▶  $-\gamma m_i \mathbf{v}_i$  is the frictional force modeling the viscous drag
- ▶  $\sqrt{2m_i \gamma k_B T} R_i(t)$  is the random force modeling the interaction with the heat bath

## Integrating the Langevin equation

- ▶ use the Langevin leap-frog method

$$\mathbf{v}_i(t + \epsilon/2) = \mathbf{v}_i(t - \epsilon/2)\alpha - (1 - \alpha)/(\gamma m_i) \frac{\partial U}{\partial \mathbf{r}_i}(t) + \sqrt{k_B T(1 - \alpha^2)/m_i} R_i(t)$$

$$\mathbf{r}_i(t + \epsilon) = \mathbf{r}_i(t) + \epsilon \mathbf{v}_i(t + \epsilon/2)$$

$$\alpha = e^{-\gamma \epsilon}$$

- ▶ sampled momentums follow the Boltzmann distribution of

$$p(\mathbf{v}_i) = \frac{1}{\sqrt{2\pi k_B T/m_i}} e^{-m_i \mathbf{v}_i^2 / 2k_B T}$$

- ▶ sampled configurations follow the Boltzmann distribution of

$$p(\mathbf{r}_1, \mathbf{r}_2, \dots, \mathbf{r}_N) = \frac{1}{Z} e^{-U(\mathbf{r}_1, \mathbf{r}_2, \dots, \mathbf{r}_N)/k_B T}$$